STUDY MODULE DESCRIPTION FORM						
Name of the module/subject Polymers and polimer composites				Code 1010702211010702974		
Field of	study	•	Profile of study	Year /Semester		
Chemical Technology			general academic, practical)	1/1		
Elective path/specialty			Subject offered in:	Course (compulsory, elective)		
Cycle of	Composit	es and Nanomaterials	Polish	obligatory		
Cycle of study:						
	Second-c	time				
No. of h	ours	4 -		No. of credits		
Lectur	e: 15 Classes	s: - Laboratory: 15	Project/seminars:	- <u> </u>		
status or the course in the study program (Basic, major, other) (university-wide, from another field) other university-wide						
Educatio	on areas and fields of sci	ence and art		ECTS distribution (number		
				and %)		
technical sciences				3 100%		
	l'echnical scie	ences		3 100%		
prof. dr hab. inż. Ewa Andrzejewska email: ewa.andrzejewska@put.poznan.pl tel. 616653637 Faculty of Chemical Technology ul. Berdychowo 4 60-965 Poznań						
Prere	quisites in term	is of knowledge, skills an	d social competencies:			
1	Knowledge	Knowledge of the basic principles of general, organic and physical chemistry. Knowledge of subjects taught at ?Chemical technology ? polymeric materials? lecture.				
2	Skills	Student knows and applies good equipment. He or she is able to other properly chosen sources.	d practices of laboratory work, is able to operate the scientific search for information in scientific literature, databases and			
3	Social competencies	Student is conscious of the effect	ts of engineering activity.			
Assu	mptions and obj	ectives of the course:				
To get	basic knowledge of p	olymers (chemistry, properties, ap	plications).			
	0	man and astronom to the	advantless - 1	a field of starts		
Kr at	Study Outco	mes and reference to the	educational results for	a field of study		
1 Stud	lent has a well establi	shed knowledge of synthesis, pror	perties anlication of polymers	- [K \\/\O2 K \\/\11]		
Skills		once knowledge of synthesis, prop		[1,_11,02,11,_11]		
1. Stud techno	lent has the ability of a logy - [K_U01, K_U	analyzing and interpreting of the re 10]	esults of experiments from the a	area of polymer chemistry and		
2. Stud	lent has the ability of	presenting the results of laborator	y exercises in concise and prop	per manner - [K_U06]		
Social competencies:						
 Itechnology, including environment protection - [K_K04, K_K02] Student is conscious of limitation of his knowledge and understands the need of further continuous education in area of polymer chemistry and technology - [K_101] 						
3. Stud teamw	lents can work in a tea ork [K_U04]	am and are aware of their respons	ibility for their work and respon	sibility for the results of the		
Assessment methods of study outcomes						

Written exam in the subject of polymeric materials presented at lectures, evaluation of laboratory exercises and reports.

Course description

Basic concepts (linear, branched and crosslinked polymers, molecular weight, tacticity). Basic characteristics of chain polymerization reaction: types, mechanisms, examples of polymers. Copolymerization and copolymers. Basic characteristics of step polymerization; mechanism, examples of polymers. Polymer morphology. Classification of polymeric materials (thermoplastics, thermosets, elastomers, thermoplastic elastomers). Polymer blends. Commodity, engineering and performance polymers. Thermal properties of polymers (thermal transitions, DSC measurements). Mechanical properties of polymers (tensile properties, stress-strain behavior failure, viscoelasticity, rheological models).

Basic bibliography:

1. G. Odian, Principles of Polymerization, 4th ed., Wiley, 2004

2. H.R. Allcock, F.W. Lampe Contemporary Polymer Chemistry, 2nd ed., Prentice Hall, 1990.

Additional bibliography:

1. L.H. Sperling Introduction to Physical Polymer Science, 4th ed., Wiley, 2006

2. Handbook of Plastics Technologies, C.A. Harper. Ed., 2006, e-book.

Result of average student's workload

Activity	Time (working hours)					
1. Lecture	15					
2. Consultations to lecture	10					
3. Laboratory	15					
4. Consultations to laboratory	10					
5. Preparation for laboratory	20					
6. Preparation of reports	5					
Student's workload						
Source of workload	hours	ECTS				
Total workload	75	3				
Contact hours	50	0				
Practical activities	15	0				